

Chapter 7 Worksheet 1 Balancing Chemical Equations Answers

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Chapter 7 Worksheet 1 Balancing

7) Copper and sulfuric acid react to form copper (II) sulfate and water and sulfur dioxide. 8) Hydrogen gas and nitrogen monoxide react to form water and nitrogen gas. Balancing Chemical Equations Answer Key

Chapter 7 Worksheet #1 Balancing Chemical Equations

Chapter 7 Worksheet #1 Balancing Chemical Equations

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7) $1 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow 1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$ 8) $1 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$ 9) $2 \text{ C}_8\text{H}_{18} + 25 \text{ O}_2 \rightarrow 16 \text{ CO}_2 + 18 \text{ H}_2\text{O}$ 10) $1 \text{ FeCl}_3 + 3 \text{ NaOH} \rightarrow 1 \text{ Fe(OH)}_3 + 3 \text{ NaCl}$ 11) $4 \text{ P} + 5 \text{ O}_2 \rightarrow 2 \text{ P}_2\text{O}_5$ 12) $2 \text{ Na} + 2 \text{ H}_2\text{O} \rightarrow 2 \text{ NaOH} + 1 \text{ H}_2$ 13) $2 \text{ Ag}_2\text{O} \rightarrow 4 \text{ Ag} + 1 \text{ O}_2$ 14) $1 \text{ S}_8 + 12 \text{ O}_2 \rightarrow 8 \text{ SO}_3$ 15) $6 \text{ CO}_2 + 6 \text{ H}_2\text{O} \rightarrow 1 \text{ C}_6\text{H}_{12}\text{O}_6 + 6 \text{ O}_2$ 16) $2 \text{ K} + 1 \text{ MgBr}_2 \rightarrow 2 \text{ KBr} + 1 \text{ Mg}$ 17)

Chapter 7 Worksheet #1 Balancing Chemical Equations ...

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Balancing Chemical Equations - Answer Key Balance the equations below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$ 5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$ 6) $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$ 7) $1 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow 1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$ 8) $1 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$ 9) $2 \text{ C}_8\text{H}_{18} + 25 \text{ O}_2 \rightarrow 16 \text{ CO}_2 + 18 \text{ H}_2\text{O}$ 10) $1 \text{ FeCl}_3 + 3 \text{ NaOH} \rightarrow 1 \text{ Fe(OH)}_3 + 3 \text{ NaCl}$ 11) $4 \text{ P} + 5 \text{ O}_2 \rightarrow 2 \text{ P}_2\text{O}_5$

H SO NaNO HNO Na SO

Balancing Equations Worksheet #1 - ANSWERS Balance the following equations by placing the correct coefficients in the space provided. 1. $\underline{2} \text{ H}_2 + \underline{1} \text{ O}_2 \rightarrow \underline{2} \text{ H}_2\text{O}$

Balancing Equations Worksheet #1 - ANSWERS

Balancing Chemical Equations Worksheet 1 - ANSWERS 1. $2 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O}$ 2. $2 \text{ Na} + \text{Cl}_2 \rightarrow 2 \text{ NaCl}$ 3. $\text{N}_2\text{O}_4 \rightarrow 2 \text{ NO}_2$ 4. $2 \text{ Mg} + \text{O}_2 \rightarrow 2 \text{ MgO}$ 5. $2 \text{ H}_2\text{O}_2 \rightarrow 2 \text{ H}_2\text{O} + \text{O}_2$ 6. $3 \text{ Ca} + \text{N}_2 \rightarrow \text{Ca}_3\text{N}_2$ 7. $2 \text{ Li} + \text{F}_2 \rightarrow 2 \text{ LiF}$ 8. $3 \text{ Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$ 9. $2 \text{ NH}_3 \rightarrow 3 \text{ N}_2 + 2 \text{ H}_2$ 10. $2 \text{ HCl} \rightarrow \text{H}_2 + \text{Cl}_2$ 11. $2 \text{ NI}_3 \rightarrow \text{N}_2 + 3 \text{ I}_2$ 12. $2 \text{ HI} \rightarrow \text{H}_2 + \text{I}_2$ 13 ...

Balancing Chemical Equations Worksheet 1

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Chapter 7 Test Worksheets - Kiddy Math

Write the word equations below as chemical equations and balance: 1) Zinc and lead (II) nitrate react to form zinc nitrate and lead. 2) Aluminum bromide and chlorine gas react to form aluminum chloride and bromine gas. 3) Sodium phosphate and calcium chloride react to form calcium phosphate and sodium chloride.

Chapter 7 Worksheet #1 Balancing Chemical Equations

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Chapter Worksheets - Lesson Worksheets

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Balancing Equations Worksheets

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